

ContainerPower Energy Solutions

Using lead-acid batteries for energy storage



Overview

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Energy storage using batteries is accepted as one of the most important and efficient ways of stabilising electricity networks and there are a variety of different battery chemistries that may be used. Lead batteries are very well established both for automotive and industrial applications and have.

A lead acid battery is a kind of rechargeable battery that stores electrical energy by using chemical reactions between lead, water, and sulfuric acid. The technology behind these batteries is over 160 years old, but the reason they're still so popular is because they're robust, reliable, and cheap.

The lead-acid (PbA) battery was invented by Gaston Planté more than 160 years ago and it was the first ever rechargeable battery. In the charged state, the positive electrode is lead dioxide (PbO₂) and the negative electrode is metallic lead (Pb); upon discharge in the sulfuric acid electrolyte.

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